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(54) Title: FUEL COMPOSITION FOR TWO-CYCLE ENGINES (57) Abstract A fuel composition for two-cycle engines comprising a major amount of fuel boiling in the gasoline range and a minor amount of a lubricant composition comprising a base oil of lubricating viscosity and an additive formulation comprising (1) a molybdenum/sulfur complex of a basic nitrogen compound, (2) a carboxylic acid amide, and (3) a succinimide.		

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01 FUEL COMPOSITION FOR TWO-CYCLE ENGINES

02

03 BACKGROUND OF THE INVENTION

04

05 The present invention relates to a fuel composition for
06 two-cycle internal combustion engines which comprises a
07 major amount of fuel boiling in the gasoline range and a
08 minor amount of a lubricant composition comprising a
09 lubricating oil and an additive formulation containing a
10 molybdenum/sulfur complex of a basic nitrogen compound.

11

12 Over the past several decades the use of spark-ignited
13 two-cycle (two-stroke) internal combustion engines including
14 rotary engines such as those of the Wankel type has steadily
15 increased. They are presently found in power lawn mowers
16 and other power-operated garden equipment, power chain saws,
17 pumps, electrical generators, marine outboard engines,
18 snowmobiles, motorcycles, and the like.

19

20 The increasing use of two-cycle engines coupled with
21 increasing severity of the conditions in which they have
22 operated has led to an increasing demand for oils to
23 adequately lubricate such engines. Among the problems
24 associated with lubrication of two-cycle engines are piston
25 ring sticking, rusting, lubrication failure of connecting
26 rods and main bearings and the general formation on the
27 engine's interior surface of carbon and varnish deposits.
28 The formation of varnish is a particularly vexatious problem
29 since the build-up of varnish on piston and cylinder walls
30 is believed to ultimately result in ring sticking which
31 leads to failure of the sealing function of piston rings.
32 Such seal failure causes loss of cylinder compression which
33 is particularly damaging in two-cycle engines because they
34 depend on suction to draw the new fuel charge into the

01 exhausted cylinder. Thus, ring sticking can lead to
02 deterioration in engine performance, and unnecessary
03 consumption of fuel and/or lubricant. Spark plug fouling
04 and engine port plugging problems also occur in two-cycle
05 engines.

06

07 A variety of compounds have been proposed as additives for
08 fuel-lubricating oil mixtures to be used in two-cycle
09 internal combustion engines. For example, U.S. Patent
10 No. 4,708,809 to Davis discloses a lubricant composition for
11 two-cycle engines comprising a major amount of an oil of
12 lubricating viscosity and a minor amount of at least one
13 alkyl phenol having at least one hydrocarbon-based group of
14 at least 10 aliphatic carbon atoms. Preferably, such
15 lubricant composition will also contain a detergent-
16 dispersant additive selected from (i) a neutral or basic
17 metal salt of an organic sulfur acid, phenol or carboxylic
18 acid, (ii) a hydrocarbyl-substituted amine, (iii) an
19 acylated, nitrogen-containing compound having a substituent
20 of at least 10 aliphatic carbon atoms, (iv) a
21 nitrogen-containing condensate of a phenol, aldehyde and
22 amino compound, and (v) an ester of a substituted
23 polycarboxylic acid.

24

25 U.S. Patent No. 4,724,091 to Davis discloses a lubricant
26 composition for two-cycle engines comprising a major amount
27 of an oil of lubricating viscosity and a minor amount of a
28 mixture of at least one alkyl phenol and at least one amino
29 phenol, each phenol having at least one hydrocarbon-based
30 group of at least about 10 aliphatic carbon atoms.
31 Preferably, this composition will additionally contain a
32 detergent-dispersant additive.

33

34

01 U.S. Patent No. 4,740,321 to Davis et al. discloses a
02 lubricant composition for two-cycle engines comprising a
03 major amount of an oil of lubricating viscosity and a minor
04 amount of at least one sulfurized alkyl phenol or metal salt
05 thereof having at least one hydrocarbon-based group of at
06 least 10 aliphatic carbon atoms. This lubricant composition
07 will also preferably contain a detergent-dispersant
08 additive.

09
10 U.S. Patent No. 4,705,643 to Nemo discloses a lubricating
11 oil composition for two-cycle engines comprising a
12 lubricating oil and an ashless detergent additive which is
13 the hydrolyzed reaction product of an aliphatic branched
14 chain carboxylic acid of 16 to 20 carbon atoms and a
15 polyamine of at least 3 amine groups. Preferably, the
16 ashless detergent additive is the hydrolyzed reaction
17 product of isostearic acid and tetraethylenepentamine.

18
19 U.S. Patent No. 4,994,196 to Kagaya et al. discloses a
20 two-cycle engine oil composition comprising a base oil and a
21 calcium phenate detergent additive, wherein the base oil is
22 a mixture of (a) a copolymer of an alpha-olefin with an
23 ester of a dicarboxylic acid and (b) an ester of
24 pentaerythritol and a fatty acid.

25
26 U.S. Patent No. 3,888,776 to Silverstein discloses a
27 two-cycle engine lubricant which comprises a major amount of
28 a polypropylene glycol and minor amounts of a sulfurized
29 oxymolybdenum organophosphorodithioate, a finely divided
30 molybdenum disulfide and a halogenated hydrocarbon
31 detergent, such as 1,1,1-trichloroethylene,
32 orthodichlorobenzene, perchlorinated biphenyl, and the like.

33
34

01 Molybdenum/sulfur complexes of basic nitrogen compounds have
02 previously been described in the art as useful antioxidant
03 additives for lubricant compositions finding application,
04 for example, as crosshead diesel engine lubricants,
05 automobile and railroad crankcase lubricants, lubricants for
06 heavy machinery, greases for bearings, and the like.
07

08 For example, U.S. Patent No. 4,263,152 to King et al.
09 discloses an antioxidant additive for lubricating oils which
10 is prepared by combining an acidic molybdenum compound, a
11 polar promoter, a basic nitrogen-containing compound and a
12 sulfur source to form a molybdenum and sulfur-containing
13 complex. Similar molybdenum-containing antioxidant
14 additives are disclosed in U.S. Patent Nos. 4,285,822;
15 4,283,295; 4,272,387; 4,265,773; 4,261,843; 4,259,195; and
16 4,259,194. However, none of these patents teaches or
17 appreciates the use of such antioxidant additives, or
18 lubricating oils containing such additives, in admixture
19 with fuels in two-cycle engines. Furthermore, none of these
20 patents teaches or appreciates that such antioxidant
21 additives would be effective deposit control agents or would
22 reduce piston sticking when utilized in fuel-lubricating oil
23 mixtures in two-cycle engines.
24

25 Moreover, as taught in the aforementioned U.S. Patent
26 No. 4,708,809, the unique problems and techniques associated
27 with the lubrication of two-cycle engines has led to the
28 recognition by those skilled in the art of two-cycle engine
29 lubricants as a distinct lubricant type.
30

31 Accordingly, the present invention is directed to minimizing
32 the problems of varnish build-up and ring sticking in
33 two-cycle engines through the provision of effective
34 additives for fuel-lubricating oil combinations which

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01 eliminate or reduce two-cycle engine varnish deposits and
02 piston ring seal failure.

03

04

SUMMARY OF THE INVENTION

05

06 The present invention provides a fuel composition for
07 two-cycle engines comprising a major amount of fuel boiling
08 in the gasoline range and a minor amount of a lubricant
09 composition comprising:

10

11 (A) a major amount of a base oil of lubricating viscosity,
12 and

13

14 (B) a minor amount of an additive formulation comprising:

15

16 (1) a sulfurized molybdenum-containing composition
17 prepared by (i) reacting an acidic molybdenum
18 compound and a basic nitrogen compound selected
19 from the group consisting of a succinimide, a
20 carboxylic acid amide, a hydrocarbyl monoamine, a
21 hydrocarbyl polyamine, a Mannich base, a
22 phosphoramidate, a thiophosphoramidate, a
23 phosphonamidate, a dispersant viscosity index
24 improver, or a mixture thereof, in the presence of
25 a polar promoter, to form a molybdenum complex
26 wherein from 0.01 to 2 atoms of molybdenum are
27 present per basic nitrogen atom, and the promoter
28 is present in the ratio of 0.01 to 50 moles of
29 polar promoter per mole of molybdenum; and
30 (ii) reacting the molybdenum complex with a
31 sulfur-containing compound in an amount sufficient
32 to provide about 1.5 to 4.0 atoms of sulfur per
33 atom of molybdenum, to thereby form a sulfur- and
34 molybdenum-containing composition,

01 (2) a carboxylic acid amide, and

02

03 (3) a succinimide.

04

05 Among other factors, the present invention is based upon the
06 unexpected discovery that additive formulations containing a
07 molybdenum/sulfur complex of a basic nitrogen compound, plus
08 a carboxylic acid amide and a succinimide are surprisingly
09 effective agents for deposit control and reduction of piston
10 ring sticking when combined in fuel-lubricating oil mixtures
11 in two-cycle engines.

12

13 DETAILED DESCRIPTION OF THE INVENTION

14

15 The fuel composition of the present invention will comprise
16 a major amount of fuel boiling in the gasoline range and a
17 minor amount of a lubricant composition comprising a base
18 oil of lubricating viscosity and an additive formulation
19 containing (1) a sulfurized molybdenum-containing
20 composition, (2) a carboxylic acid amide, and (3) a
21 succinimide.

22

23 The sulfurized molybdenum-containing composition employed in
24 the present invention may be generally characterized as a
25 molybdenum/sulfur complex of a basic nitrogen compound.
26 Such molybdenum/sulfur complexes are known in the art and
27 are described, for example, in U.S. Patent No. 4,263,152 to
28 King et al., the disclosure of which is hereby incorporated
29 by reference.

30

31 The precise molecular formula of the molybdenum compositions
32 employed in this invention is not known with certainty;
33 however, they are believed to be compounds in which
34 molybdenum, whose valences are satisfied with atoms of

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01 oxygen or sulfur, is either complexed by, or the salt of,
02 one or more nitrogen atoms of the basic nitrogen containing
03 compound used in the preparation of these compositions.
04

05 The molybdenum compounds used to prepare the
06 molybdenum/sulfur complexes employed in this invention are
07 acidic molybdenum compounds. By acidic is meant that the
08 molybdenum compounds will react with a basic nitrogen
09 compound as measured by ASTM test D-664 or D-2896 titration
10 procedure. Typically these molybdenum compounds are
11 hexavalent and are represented by the following
12 compositions: molybdic acid, ammonium molybdate, sodium
13 molybdate, potassium molybdate and other alkaline metal
14 molybdates and other molybdenum salts such as hydrogen
15 salts, e.g., hydrogen sodium molybdate, MoOCl_4 , MoO_2Br_2 ,
16 $\text{Mo}_2\text{O}_3\text{Cl}_6$, molybdenum trioxide or similar acidic molybdenum
17 compounds. Preferred acidic molybdenum compounds are
18 molybdic acid, ammonium molybdate, and alkali metal
19 molybdates. Particularly preferred are molybdic acid and
20 ammonium molybdate.
21

22 The basic nitrogen compound used to prepare the
23 molybdenum/sulfur complexes must have a basic nitrogen
24 content as measured by ASTM D-664 or D-2896. It is
25 preferably oil-soluble. Typical of such compositions are
26 succinimides, carboxylic acid amides, hydrocarbyl
27 monoamines, hydrocarbon polyamines, Mannich bases,
28 phosphoramides, thiophosphoramides, phosphonamides,
29 dispersant viscosity index improvers, and mixtures thereof.
30 These basic nitrogen-containing compounds are described
31 below (keeping in mind the reservation that each must have
32 at least one basic nitrogen). Any of the
33 nitrogen-containing compositions may be after-treated with,
34

01 e.g., boron, using procedures well known in the art so long
02 as the compositions continue to contain basic nitrogen.
03 These after-treatments are particularly applicable to
04 succinimides and Mannich base compositions.
05

06 The mono and polysuccinimides that can be used to prepare
07 the molybdenum/sulfur complexes described herein are
08 disclosed in numerous references and are well known in the
09 art. Certain fundamental types of succinimides and the
10 related materials encompassed by the term of art
11 "succinimide" are taught in U.S. Patent Nos. 3,219,666;
12 3,172,892; and 3,272,746, the disclosures of which are
13 hereby incorporated by reference. The term "succinimide" is
14 understood in the art to include many of the amide, imide,
15 and amidine species which may also be formed. The
16 predominant product however is a succinimide and this term
17 has been generally accepted as meaning the product of a
18 reaction of an alkenyl substituted succinic acid or
19 anhydride with a nitrogen-containing compound. Preferred
20 succinimides, because of their commercial availability, are
21 those succinimides prepared from a hydrocarbyl succinic
22 anhydride, wherein the hydrocarbyl group contains from about
23 24 to about 350 carbon atoms, and an ethylene amine, said
24 ethylene amines being especially characterized by ethylene
25 diamine, diethylene triamine, triethylene tetramine, and
26 tetraethylene pentamine. Particularly preferred are those
27 succinimides prepared from polyisobutenyl succinic anhydride
28 of 70 to 128 carbon atoms and tetraethylene pentamine or
29 triethylene tetramine or mixtures thereof.
30

31 Also included within the term "succinimide" are the
32 cooligomers of a hydrocarbyl succinic acid or anhydride and
33 a poly secondary amine containing at least one tertiary
34 amino nitrogen in addition to two or more secondary amino

01 groups. Ordinarily this composition has between 1,500 and
02 50,000 average molecular weight. A typical compound would
03 be that prepared by reacting polyisobutenyl succinic
04 anhydride and ethylene dipiperazine.

05

06 Carboxylic acid amide compositions are also suitable
07 starting materials for preparing the molybdenum/sulfur
08 complexes employed in this invention. Typical of such
09 compounds are those disclosed in U.S. Patent No. 3,405,064,
10 the disclosure of which is hereby incorporated by reference.
11 These compositions are ordinarily prepared by reacting a
12 carboxylic acid or anhydride or ester thereof, having at
13 least 12 to about 350 aliphatic carbon atoms in the
14 principal aliphatic chain and, if desired, having sufficient
15 pendant aliphatic groups to render the molecule oil soluble
16 with an amine or a hydrocarbyl polyamine, such as an
17 ethylene amine, to give a mono or polycarboxylic acid amide.
18 Preferred are those amides prepared from (1) a carboxylic
19 acid of the formula R^2COOH , where R^2 is C_{12-20} alkyl or a
20 mixture of this acid with a polyisobutenyl carboxylic acid
21 in which the polyisobutenyl group contains from 72 to
22 128 carbon atoms and (2) an ethylene amine, especially
23 triethylene tetramine or tetraethylene pentamine or mixtures
24 thereof.

25

26 Another class of compounds which are useful in this
27 invention are hydrocarbyl monoamines and hydrocarbyl
28 polyamines, preferably of the type disclosed in U.S. Patent
29 No. 3,574,576, the disclosure of which is hereby
30 incorporated by reference. The hydrocarbyl group, which is
31 preferably alkyl, or olefinic having one or two sites of
32 unsaturation, usually contains from 9 to 350, preferably
33 from 20 to 200 carbon atoms. Particularly preferred
34

01 hydrocarbyl polyamines are those which are derived, e.g., by
02 reacting polyisobutenyl chloride and a polyalkylene
03 polyamine, such as an ethylene amine, e.g., ethylene
04 diamine, diethylene triamine, tetraethylene pentamine,
05 2-aminoethylpiperazine, 1,3-propylene diamine,
06 1,2-propylenediamine, and the like.

07
08 Another class of compounds useful for supplying basic
09 nitrogen are the Mannich base compositions. These
10 compositions are prepared from a phenol or C₉₋₂₀₀
11 alkylphenol, an aldehyde, such as formaldehyde or
12 formaldehyde precursor such as paraformaldehyde, and an
13 amine compound. The amine may be a mono or polyamine and
14 typical compositions are prepared from an alkylamine, such
15 as methylamine or an ethylene amine, such as, diethylene
16 triamine, or tetraethylene pentamine, and the like. The
17 phenolic material may be sulfurized and preferably is
18 dodecylphenol or a C₈₀₋₁₀₀ alkylphenol. Typical Mannich
19 bases which can be used in this invention are disclosed in
20 U.S. Patent No. 4,157,309 and U.S. Patent Nos. 3,649,229;
21 3,368,972; and 3,539,663, the disclosures of which are
22 hereby incorporated by reference. The last referenced
23 patent discloses Mannich bases prepared by reacting an
24 alkylphenol having at least 50 carbon atoms, preferably 50
25 to 200 carbon atoms with formaldehyde and an alkylene
26 polyamine $\text{HN}(\text{ANH})_n\text{H}$ where A is a saturated divalent alkyl
27 hydrocarbon of 2 to 6 carbon atoms and n is 1-10 and where
28 the condensation product of said alkylene polyamine may be
29 further reacted with urea or thiourea. The utility of these
30 Mannich bases as starting materials for preparing
31 lubricating oil additives can often be significantly
32 improved by treating the Mannich base using conventional
33 techniques to introduce boron into the composition.
34

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01 Another class of composition useful for preparing the
02 molybdenum/sulfur complexes employed in this invention are
03 the phosphoramides and phosphonamides such as those
04 disclosed in U.S. Patent Nos. 3,909,430 and 3,968,157, the
05 disclosures of which are hereby incorporated by reference.
06 These compositions may be prepared by forming a phosphorus
07 compound having at least one P-N bond. They can be
08 prepared, for example, by reacting phosphorus oxychloride
09 with a hydrocarbyl diol in the presence of a monoamine or by
10 reacting phosphorus oxychloride with a difunctional
11 secondary amine and a mono-functional amine.

12 Thiophosphoramides can be prepared by reacting an
13 unsaturated hydrocarbon compound containing from 2 to 450 or
14 more carbon atoms, such as polyethylene, polyisobutylene,
15 polypropylene, ethylene, 1-hexene, 1,3-hexadiene,
16 isobutylene, 4-methyl-1-pentene, and the like, with
17 phosphorus pentasulfide and a nitrogen-containing compound
18 as defined above, particularly an alkylamine, alkyldiamine,
19 alkylpolyamine, or an alkyleneamine, such as ethylene
20 diamine, diethylenetriamine, triethylenetetramine,
21 tetraethylenepentamine, and the like.

22
23 Another class of nitrogen-containing compositions useful in
24 preparing the molybdenum complexes employed in this
25 invention includes the so-called dispersant viscosity index
26 improvers (VI improvers). These VI improvers are commonly
27 prepared by functionalizing a hydrocarbon polymer,
28 especially a polymer derived from ethylene and/or propylene,
29 optionally containing additional units derived from one or
30 more co-monomers such as alicyclic or aliphatic olefins or
31 diolefins. The functionalization may be carried out by a
32 variety of processes which introduce a reactive site or
33 sites which usually has at least one oxygen atom on the
34 polymer. The polymer is then contacted with a

01 nitrogen-containing source to introduce nitrogen-containing
02 functional groups on the polymer backbone. Commonly used
03 nitrogen sources include any basic nitrogen compound
04 especially those nitrogen-containing compounds and
05 compositions described herein. Preferred nitrogen sources
06 are alkylene amines, such as ethylene amines, alkyl amines,
07 and Mannich bases.
08

09 Preferred basic nitrogen compounds for use in this invention
10 are succinimides, carboxylic acid amides, and Mannich bases.
11

12 Representative sulfur sources for preparing the molybdenum
13 complexes used in this invention are sulfur, hydrogen
14 sulfide, sulfur monochloride, sulfur dichloride, phosphorus
15 pentasulfide, R_2S_x where R is hydrocarbyl, preferably C_{1-40}
16 alkyl, and x is at least 2, inorganic sulfides and
17 polysulfides such as $(NH_4)_2S_x$, where x is at least 1,
18 thioacetamide, thiourea, and mercaptans of the formula RSH
19 where R is as defined above. Also useful as sulfurizing
20 agents are traditional sulfur-containing antioxidants such
21 as wax sulfides and polysulfides, sulfurized olefins,
22 sulfurized carboxylic and esters and sulfurized
23 ester-olefins, and sulfurized alkylphenols and the metal
24 salts thereof.
25

26 The sulfurized fatty acid esters are prepared by reacting
27 sulfur, sulfur monochloride, and/or sulfur dichloride with
28 an unsaturated fatty ester under elevated temperatures.
29 Typical esters include C_1-C_{20} alkyl esters of C_8-C_{24}
30 unsaturated fatty acids, such as palmitoleic, oleic,
31 ricinoleic, petroselinic, vaccenic, linoleic, linolenic,
32 oleostearic, licanic, paranaric, tariric, gadoleic,
33 arachidonic, cetoleic, etc. Particularly good results have
34

01 been obtained with mixed unsaturated fatty acid esters, such
02 as are obtained from animal fats and vegetable oils, such as
03 tall oil, linseed oil, olive oil, castor oil, peanut oil,
04 rape oil, fish oil, sperm oil, and so forth.

05

06 Exemplary fatty esters include lauryl tallate, methyl
07 oleate, ethyl oleate, lauryl oleate, cetyl oleate, cetyl
08 linoleate, lauryl ricinoleate, oleyl linoleate, oleyl
09 stearate, and alkyl glycerides.

10

11 Cross-sulfurized ester olefins, such as a sulfurized mixture
12 of C_{10} - C_{25} olefins with fatty acid esters of C_{10} - C_{25} fatty
13 acids and C_1 - C_{25} alkyl or alkenyl alcohols, wherein the
14 fatty acid and/or the alcohol is unsaturated may also be
15 used.

16

17 Sulfurized olefins are prepared by the reaction of the C_3 - C_6
18 olefin or a low-molecular-weight polyolefin derived
19 therefrom with a sulfur-containing compound such as sulfur,
20 sulfur monochloride, and/or sulfur dichloride.

21

22 Also useful are the aromatic and alkyl sulfides, such as
23 dibenzyl sulfide, dixylyl sulfide, dicetyl sulfide,
24 diparaffin wax sulfide and polysulfide, cracked wax-olefin
25 sulfides and so forth. They can be prepared by treating the
26 starting material, e.g., olefinically unsaturated compounds,
27 with sulfur, sulfur monochloride, and sulfur dichloride.
28 Particularly preferred are the paraffin wax thiomers
29 described in U.S. Patent No. 2,346,156.

30

31 Sulfurized alkyl phenols and the metal salts thereof include
32 compositions such as sulfurized dodecylphenol and the
33 calcium salts thereof. The alkyl group ordinarily contains
34

01 from 9-300 carbon atoms. The metal salt may be preferably,
02 a Group I or Group II salt, especially sodium, calcium,
03 magnesium, or barium.

04

05 Preferred sulfur sources are sulfur, hydrogen sulfide,
06 phosphorus pentasulfide, R_2S_x where R is hydrocarbyl,
07 preferably C_1 - C_{10} alkyl, and x is at least 3, mercaptans
08 wherein R is C_1 - C_{10} alkyl, inorganic sulfides and
09 polysulfides, thioacetamide, and thiourea. Most preferred
10 sulfur sources are sulfur, hydrogen sulfide, phosphorus
11 pentasulfide, and inorganic sulfides and polysulfides.
12

13 The polar promoter used in the preparation of the molybdenum
14 complexes employed in this invention is one which
15 facilitates the interaction between the acidic molybdenum
16 compound and the basic nitrogen compound. A wide variety of
17 such promoters are well known to those skilled in the art.
18 Typical promoters are 1,3-propanediol, 1,4-butane-diol,
19 diethylene glycol, butyl cellosolve, propylene glycol,
20 1,4-butyleneglycol, methyl carbitol, ethanolamine,
21 diethanolamine, N-methyl-diethanol-amine, dimethyl
22 formamide, N-methyl acetamide, dimethyl acetamide, methanol,
23 ethylene glycol, dimethyl sulfoxide, hexamethyl
24 phosphoramide, tetrahydrofuran and water. Preferred are
25 water and ethylene glycol. Particularly preferred is water.
26

27 While ordinarily the polar promoter is separately added to
28 the reaction mixture, it may also be present, particularly
29 in the case of water, as a component of non-anhydrous
30 starting materials or as waters of hydration in the acidic
31 molybdenum compound, such as $(NH_4)_6Mo_7O_{24} \cdot 4 H_2O$. Water may
32 also be added as ammonium hydroxide.
33

34

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01 A method for preparing the molybdenum/sulfur complexes used
02 in this invention is to prepare a solution of the acidic
03 molybdenum precursor and a polar promoter with a basic
04 nitrogen-containing compound with or without diluent. The
05 diluent is used, if necessary, to provide a suitable
06 viscosity for easy stirring. Typical diluents are
07 lubricating oil and liquid compounds containing only carbon
08 and hydrogen. If desired, ammonium hydroxide may also be
09 added to the reaction mixture to provide a solution of
10 ammonium molybdate. This reaction is carried out at a
11 temperature from the melting point of the mixture to reflux
12 temperature. It is ordinarily carried out at atmospheric
13 pressure although higher or lower pressures may be used if
14 desired. This reaction mixture is treated with a sulfur
15 source as defined above at a suitable pressure and
16 temperature for the sulfur source to react with the acidic
17 molybdenum and basic nitrogen compounds. In some cases,
18 removal of water from the reaction mixture may be desirable
19 prior to completion of reaction with the sulfur source.
20

21 In the reaction mixture, the ratio of molybdenum compound to
22 basic nitrogen compound is not critical; however, as the
23 amount of molybdenum with respect to basic nitrogen
24 increases, the filtration of the product becomes more
25 difficult. Since the molybdenum component probably
26 oligomerizes, it is advantageous to add as much molybdenum
27 as can easily be maintained in the composition. Usually,
28 the reaction mixture will have charged to it from 0.01 to
29 2.00 atoms of molybdenum per basic nitrogen atom.
30 Preferably from 0.4 to 1.0, and most preferably from 0.4 to
31 0.7, atoms of molybdenum per atom of basic nitrogen is added
32 to the reaction mixture.
33
34

01 The sulfur source is usually charged to the reaction mixture
02 in such a ratio to provide 1.5 to 4.0 atoms of sulfur per
03 atom of molybdenum. Preferably from 2.0 to 4.0 atoms of
04 sulfur per atom of molybdenum is added, and most preferably,
05 2.5 to 4.0 atoms of sulfur per atom of molybdenum.
06

07 The polar promoter, which is preferably water, is ordinarily
08 present in the ratio of 0.1 to 50 moles of promoter per mole
09 of molybdenum. Preferably from 0.5 to 25 and most
10 preferably 1.0 to 15 moles of the promoter is present per
11 mole of molybdenum.
12

13 As described above, the additive formulation employed in the
14 present invention contains (1) a sulfurized
15 molybdenum-containing composition, (2) a carboxylic acid
16 amide, and (3) a succinimide.
17

18 The carboxylic amide component of the presently employed
19 additive formulation may be any of the carboxylic acid amide
20 compounds described herein as useful in the preparation of
21 the molybdenum/sulfur complex. Preferred carboxylic acid
22 amide components include those amides derived from a
23 carboxylic acid of the formula $R^2\text{COOH}$, wherein R^2 is $\text{C}_{12}\text{-C}_{20}$
24 alkyl, and an ethylene amine, such as triethylene tetramine
25 or tetraethylene pentamine.
26

27 Similarly, the succinimide component of the presently
28 employed additive formulation may be any of the succinimide
29 compounds described herein as useful in the preparation of
30 the molybdenum/sulfur complex. Preferred succinimide
31 components include those derived from polyisobutenyl
32 succinic anhydride, wherein the polyisobutenyl group
33 contains from about 50 to 250 carbon atoms, and an ethylene
34

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01 amine, such as triethylene tetramine or tetraethylene
02 pentamine.

03

04 The additive formulation employed in the present invention
05 may additionally contain a flocculant inhibitor and/or a
06 lubricity agent, such as a polyisobutene. If necessary, a
07 diluent oil may also be included.

08

09 Other additives such as viscosity index improvers,
10 antioxidants, dispersants, coupling agents, pour point
11 depressants, extreme pressure agents, color stabilizers,
12 rust inhibitors, anticorrosion agents, and the like, may
13 also be present in the additive formulation.

14

15 The lubricant composition employed in the present invention
16 comprises a major amount of a base oil of lubricating
17 viscosity and a minor amount of the additive formulation
18 described above.

19

20 The base oil employed may be any of a wide variety of oils
21 of lubricating viscosity. Thus, the base oil can be a
22 refined paraffin type base oil, a refined naphthenic base
23 oil, or a synthetic hydrocarbon or non-hydrocarbon oil of
24 lubricating viscosity. The base oil can also be a mixture
25 of mineral and synthetic oils. For purposes of the present
26 invention, the mineral lubricating oils are preferred, since
27 they are presently in more general use in two-cycle engines.

28

29 The presently employed lubricant composition containing the
30 additive formulation described herein can be conveniently
31 prepared using conventional techniques by admixing the
32 appropriate amount of each component of the additive
33 formulation with a lubricating oil.

34

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01 Generally, the amount of the molybdenum-containing additive
02 will vary from about 0.05 to 15% by weight and preferably
03 from about 0.2 to 10% by weight, based on the total
04 lubricant composition, including base oil. The carboxylic
05 acid amide component will vary from about 0.05 to 20% by
06 weight and preferably from about 0.2 to 15% by weight. The
07 succinimide component will vary from about 0.5 to 15% by
08 weight and preferably from about 0.2 to 10% by weight.

09

10 The two-cycle engine fuel composition contemplated by the
11 present invention comprises a major amount of fuel boiling
12 in the gasoline range and minor amount of the lubricant
13 composition disclosed herein.

14

15 For purposes of the present invention, the lubricant
16 composition will generally be added directly to the fuel to
17 form a mixture of lubricant and fuel which is then
18 introduced into the two-cycle engine cylinder. Generally,
19 the resulting fuel composition will contain from about 15 to
20 250 parts fuel per 1 part lubricant, and more typically
21 about 50 to 100 parts fuel per 1 part lubricant. For some
22 two-cycle engine applications, the lubricant may be directly
23 injected into the combustion chamber along with the fuel or
24 into the fuel just prior to the time the fuel enters the
25 combustion chamber.

26

27 The fuel employed in the present fuel composition is a
28 hydrocarbon distillate fuel boiling in the gasoline range.
29 In such gasoline fuels, other fuel additives may also be
30 included such as antiknock agents, e.g.,
31 methylcyclopentadienyl manganese tricarbonyl, tetramethyl or
32 tetraethyl lead, or other dispersants or detergents such as
33 various substituted amines, etc. Also included may be lead
34 scavengers such as aryl halides, e.g., dichlorobenzene or

01 alkyl halides, e.g., ethylene dibromide. Additionally,
02 antioxidants, metal deactivators, pour point depressants,
03 corrosion inhibitors and demulsifiers may be present.
04

05 The following examples are presented to illustrate specific
06 embodiments of this invention and are not to be construed in
07 any way as limiting the scope of the invention.
08

09 EXAMPLES

10
11 Example 1
12

13 To a 5000 ml flask was added 114 grams molybdenum trioxide
14 and 196 grams of water. Stirring was started and 1200 grams
15 of a solution of a 45% concentrate in oil of the
16 polyisobutenyl succinimide prepared from polyisobutenyl
17 succinic anhydride having a number average molecular weight
18 for the polyisobutenyl group of about 950 and tetraethylene
19 pentamine, and 1200 grams of hydrocarbon thinner were added.
20 The mixture was refluxed at 100°C for 3 hours. The
21 temperature was gradually increased over approximately
22 1 hour to 170°C while distilling water. The temperature was
23 maintained an additional hour after the water was removed.
24 The temperature was lowered to 100°C-120°C and the mixture
25 filtered and returned to the reaction vessel. To the
26 solution was added 51 grams of sulfur. The mixture was
27 heated to 160°C-180°C for 7 hours. The pressure was slowly
28 reduced to about 50 mm of mercury to remove the hydrocarbon
29 thinner. This produced 1244 grams of product containing
30 1.80% nitrogen, 5.63% molybdenum, and 3.57% sulfur.
31
32
33
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01

Example 2

02

03 To a 5000 ml flask was added 52 grams molybdenum trioxide
04 and 111 grams of water. Stirring was started and 1184 grams
05 of a solution of a 45% concentrate in oil of the succinimide
06 described in Example 1 and 1184 grams of hydrocarbon thinner
07 were added. The mixture was refluxed at 100°C for 3 hours.
08 The temperature was gradually increased over approximately
09 1 hour to 170°C while distilling water. The temperature was
10 maintained an additional hour after the water was removed.
11 The temperature was lowered to 100°C-120°C and the mixture
12 filtered and returned to the reaction vessel. To the
13 solution was added 47 grams of sulfur. The mixture was
14 heated to 160°C-180°C for 7 hours. The pressure was slowly
15 reduced to about 50 mm of mercury to remove the hydrocarbon
16 thinner. This produced 1220 grams of product containing
17 1.94% nitrogen, 2.78% molybdenum, and 3.64% sulfur.

18

19

Example 3

20

21 To a 5000 ml flask was added 49 grams molybdenum trioxide
22 and 105 grams of water. Stirring was started and 1133 grams
23 of a solution of a 45% concentrate in oil of the succinimide
24 described in Example 1 and 1133 grams of hydrocarbon thinner
25 were added. The mixture was refluxed at 100°C for 3 hours.
26 The temperature was gradually increased over approximately
27 1 hour to 170°C while distilling water. The temperature was
28 maintained an additional hour after the water was removed.
29 The temperature was lowered to 100°C-120°C and the mixture
30 filtered and returned to the reaction vessel. To the
31 solution was added 22 grams of sulfur. The mixture was
32 heated to 160°C-180°C for 7 hours. The pressure was slowly
33 reduced to about 50 mm of mercury to remove the hydrocarbon

34

-21-

01 thinner. This produced 1163 grams of product containing
02 1.83% nitrogen, 2.79% molybdenum, and 1.97% sulfur.

03

04

Example 4

05

06 To a 5000 ml flask was added 1200 grams of a polyamide
07 prepared from a C₁₈ carboxylic acid and tetraethylene
08 pentamine and containing 6.4% nitrogen, 1200 grams
09 hydrocarbon thinner, 42 grams molybdenum trioxide, and
10 90 grams water. The mixture was refluxed at 100°C for
11 3 hours. The temperature was gradually increased over
12 approximately 1 hour to 170°C while distilling water. The
13 temperature was maintained an additional hour after the
14 water was removed. The temperature was lowered to
15 100°C-120°C and the mixture filtered and returned to the
16 reaction vessel. To the solution was added 21 grams of
17 sulfur. The mixture was heated to 160°C-180°C for 7 hours.
18 The pressure was slowly reduced to about 50 mm of mercury to
19 remove the hydrocarbon thinner. This produced a product
20 containing 5.88% nitrogen, 2.29% molybdenum, and 1.63%
21 sulfur.

22

23

Example 5

24

25 The molybdenum/sulfur complexes of Examples 1, 2 and 4 were
26 formulated to provide lubricant compositions containing 10%
27 of the carboxylic acid amide reaction product of isostearic
28 acid and tetraethylene pentamine, 2% of a polyisobutenyl
29 succinimide prepared from polyisobutenyl succinic anhydride
30 wherein the polyisobutenyl group has a number average
31 molecular weight of about 950 and tetraethylene pentamine,
32 2% of the molybdenum/sulfur complex of Examples 1, 2 and 4,
33 respectively, 5% of a polyisobutene having a number average
34

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01 molecular weight of about 950 as a lubricity agent, 0.5% of
02 a flocculant inhibitor, 1% of a diluent oil and about 79.5%
03 of a base oil. The base oil contains about 10% of a
04 150 bright stock, about 70% of a mixture of 350N and 650N
05 neutral oils, and about 20% of a petroleum distillate
06 solvent.

07
08 Example 6

09 Two-Cycle Gasoline Engine Test
10

11 This test was used to evaluate the detergency and general
12 performance of the fuel composition of this invention in a
13 two-cycle water-cooled outboard engine. Piston varnish,
14 ring sticking and general engine deposits were evaluated.
15

16 The test engine used was an Outboard Marine Company Johnson
17 Model No. J70ELEIE outboard engine, which is a
18 70 horsepower, water-cooled, three-cylinder, two-cycle
19 engine.
20

21 The test procedure involved a two-hour break-in period,
22 wherein the engine was run at 3,000 rpm for 1 hour, then at
23 4,000 rpm for 1 hour, using a fuel:lubricant ratio of 50:1.
24

25 The test was then conducted for 98 hours using a 50:1 fuel
26 to lubricant ratio on a 55 minute wide-open throttle,
27 5 minute idle cycle. The total test time, including
28 break-in, was 100 hours.
29

30 At the conclusion of the test, the engine was disassembled
31 and rated. The average piston rating and average
32 second-ring sticking rating for 3 cylinders was measured.
33 In the rating system employed, the higher the numerical
34 rating, the better the cleanliness performance, with 10.0

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01 being the maximum rating. Except for the piston rings, the
02 ratings are for cleanliness. The piston rings are rated for
03 the degree of sticking, with a rating of 10.0 indicating a
04 completely free piston ring.

05

06 The second-ring sticking values include a National Marine
07 Manufacturers Association (NMMA) rating, a visual rating,
08 and an adjusted rating, which is an average of the NMMA and
09 visual ratings.

10

11 The reference oil employed in this test was NMMA reference
12 oil TCW II, used as an industry standard in two-cycle engine
13 tests to measure engine cleanliness. The TCW II reference
14 oil is a standard mineral lubricating oil containing a
15 commercial ashless dispersant for gasoline two-cycle
16 engines. The reference oil is available from Citgo
17 Petroleum Corporation, Tulsa, Oklahoma.

18

19 Engine test runs were performed with a 50:1 fuel to
20 lubricant ratio, using lubricant compositions containing the
21 molybdenum/sulfur complexes of Examples 1, 2 and 4,
22 formulated as described in Example 5. The results of the
23 engine tests are shown in Table 1.

24

25 The results shown in Table 1 demonstrate that the fuel
26 composition of the present invention is highly effective in
27 reducing piston deposits and piston ring sticking in
28 two-cycle engines, and generally exceeds the performance of
29 a fuel containing the industry standard reference oil.

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TABLE 1							
TWO-CYCLE ENGINE TEST RESULTS							
Average Rating for 3 Cylinders							
Additive	Piston Skirt	Crown	2-Ring Land	Under-Crown	Second-Ring Sticking		
					NMMA	Visual	Adj.
TCW II Reference Oil	7.1	2.8	3.7	2.8	6.8	5.7	6.3
Example 1(a)	7.4	3.4	4.4	2.5	7.1	6.1	6.6
Example 4(a)	7.7	3.1	4.8	3.9	7.8	6.8	7.3
Example 2(a)	7.8	3.4	4.4	3.3	8.5	6.5	7.5

(a) Formulated as in Example 5.

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01 WHAT IS CLAIMED IS:

02

03 1. A fuel composition for two-cycle engines comprising a
04 major amount of fuel boiling in the gasoline range and
05 a minor amount of a lubricant composition comprising:

06

07 (A) a major amount of a base oil of lubricating
08 viscosity, and

09

10 (B) a minor amount of an additive formulation
11 comprising:

12

13 (1) a sulfurized molybdenum-containing
14 composition prepared by (i) reacting an
15 acidic molybdenum compound and a basic
16 nitrogen compound selected from the group
17 consisting of a succinimide, a carboxylic
18 acid amide, a hydrocarbyl monoamine, a
19 hydrocarbyl polyamine, a Mannich base, a
20 phosphoramidate, a thiophosphoramidate, a
21 phosphonamidate, a dispersant viscosity index
22 improver, or a mixture thereof, in the
23 presence of a polar promoter, to form a
24 molybdenum complex wherein from 0.01 to
25 2 atoms of molybdenum are present per basic
26 nitrogen atom, and the promoter is present in
27 the ratio of 0.01 to 50 moles of polar
28 promoter per mole of molybdenum; and
29 (ii) reacting the molybdenum complex with a
30 sulfur-containing compound in an amount
31 sufficient to provide about 1.5 to 4.0 atoms
32 of sulfur per atom of molybdenum, to thereby
33 form a sulfur- and molybdenum-containing
34 composition,

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- 01 (2) a carboxylic acid amide, and
02
- 03 (3) a succinimide.
04
- 05 2. The fuel composition of Claim 1, wherein the sulfur
06 source for component (1) is sulfur, hydrogen sulfide,
07 phosphorus pentasulfide, R_2S_x where R is hydrocarbyl,
08 and x is at least 2, inorganic sulfides or inorganic
09 polysulfides, thioacetamide, thiourea, mercaptans of
10 the formula RSH where R is hydrocarbyl, or a
11 sulfur-containing antioxidant.
12
- 13 3. The fuel composition of Claim 2, wherein the sulfur
14 source for component (1) is sulfur, hydrogen sulfide,
15 phosphorus pentasulfide, R_2S_x where R is C_{1-4}
16 hydrocarbyl, and x is at least 3, inorganic sulfides,
17 or inorganic polysulfides, thioacetamide, thiourea or
18 RSH where R is C_{1-40} alkyl, and the acidic molybdenum
19 compound is molybdic acid, ammonium molybdate, or
20 alkali metal molybdate.
21
- 22 4. The fuel composition of Claim 3, wherein said sulfur
23 source is sulfur, hydrogen sulfide, RSH where R is C_{1-10}
24 alkyl, phosphorus pentasulfide, or $(NH_4)_2S_x$, where x'
25 is at least 1, said acidic molybdenum compound is
26 molybdic acid, or ammonium molybdate, and said basic
27 nitrogen compound is a succinimide, carboxylic acid
28 amide, or Mannich base.
29
- 30 5. The fuel composition of Claim 4, wherein said basic
31 nitrogen compound is a C_{24-350} hydrocarbyl succinimide,
32 carboxylic acid amide, or a Mannich base prepared from
33 a C_{9-200} alkylphenol, formaldehyde, and an amine.
34

- 01 6. The fuel composition of Claim 5, wherein said basic
02 nitrogen compound is a polyisobutenyl succinimide
03 prepared from polyisobutenyl succinic anhydride and
04 tetraethylene pentamine or triethylene tetramine.
05
- 06 7. The fuel composition of Claim 5, wherein said basic
07 nitrogen compound is a carboxylic acid amide prepared
08 from one or more carboxylic acids of the formula
09 $R^2\text{COOH}$, or a derivative thereof which upon reaction
10 with an amine yields a carboxylic acid amide, wherein
11 R^2 is C_{12-350} alkyl or C_{12-350} alkenyl and a hydrocarbyl
12 polyamine.
13
- 14 8. The fuel composition of Claim 7, wherein R^2 is C_{12-20}
15 alkyl or C_{12-20} alkenyl and the hydrocarbyl polyamine is
16 tetraethylene pentamine or triethylene tetramine.
17
- 18 9. The fuel composition of Claim 5, wherein said basic
19 nitrogen compound is a Mannich base prepared from
20 dodecylphenol, formaldehyde, and methylamine.
21
- 22 10. The fuel composition of Claim 5, wherein said basic
23 nitrogen compound is a Mannich base prepared from
24 C_{80-100} alkylphenol, formaldehyde and triethylene
25 tetramine, tetraethylene pentamine, or mixtures
26 thereof.
27
- 28 11. The fuel composition of Claim 1, wherein the polar
29 promoter is water.
30
- 31 12. The fuel composition of Claim 1, wherein the carboxylic
32 acid amide of component (2) is derived from a
33
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- 01 carboxylic acid of the formula $R^2\text{COOH}$, wherein R^2 is
02 C_{12-20} alkyl, and an ethylene amine.
03
- 04 13. The fuel composition of Claim 1, wherein the
05 succinimide of component (3) is derived from
06 polyisobutenyl succinic anhydride, wherein the
07 polyisobutenyl group contains from about 50 to
08 250 carbon atoms, and an ethylene amine.
09
- 10 14. The fuel composition of Claim 1, wherein the lubricant
11 composition contains about 0.05 to 15% by weight of the
12 molybdenum-containing composition of component (1),
13 about 0.05 to 20% by weight of the carboxylic acid
14 amide of component (2), and about 0.05 to 15% by weight
15 of the succinimide of component (3).
16
- 17 15. The fuel composition of Claim 1, wherein the molybdenum
18 complex is reacted with the sulfur-containing compound
19 in an amount sufficient to provide about 2.0 to
20 4.0 atoms of sulfur per atom of molybdenum.
21
- 22 16. The fuel composition of Claim 15, wherein the
23 molybdenum complex is reacted with the
24 sulfur-containing compound in an amount sufficient to
25 provide about 2.5 to 4.0 atoms of sulfur per atom of
26 molybdenum.
27
- 28 17. The fuel composition of Claim 1, wherein the additive
29 formulation further contains a flocculant inhibitor.
30
- 31 18. The fuel composition of Claim 17, wherein the additive
32 formulation further contains a lubricity agent.
33
34

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- 01 19. A method for reducing engine deposits and piston ring
02 sticking in a two-cycle engine which comprises
03 operating the two-cycle engine with a fuel composition
04 comprising a major amount of fuel boiling in the
05 gasoline range and a minor amount of a lubricant
06 composition comprising:
07
- 08 (A) a major amount of a base oil of lubricating
09 viscosity, and
10
- 11 (B) a minor amount effective to reduce engine deposits
12 and piston ring sticking of an additive
13 formulation comprising:
14
- 15 (1) a sulfurized molybdenum-containing
16 composition prepared by (i) reacting an
17 acidic molybdenum compound and a basic
18 nitrogen compound selected from the group
19 consisting of a succinimide, a carboxylic
20 acid amide, a hydrocarbyl monoamine, a
21 hydrocarbyl polyamine, a Mannich base, a
22 phosphoramidate, a thiophosphoramidate, a
23 phosphonamide, a dispersant viscosity index
24 improver, or a mixture thereof, in the
25 presence of a polar promoter, to form a
26 molybdenum complex wherein from 0.01 to
27 2 atoms of molybdenum are present per basic
28 nitrogen atom, and the promoter is present in
29 the ratio of 0.01 to 50 moles of polar
30 promoter per mole of molybdenum; and
31 (ii) reacting the molybdenum complex with a
32 sulfur-containing compound in an amount
33 sufficient to provide about 1.5 to 4.0 atoms
34 of sulfur per atom of molybdenum, to thereby

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- 01 form a sulfur- and molybdenum-containing
02 composition;
03
04 (2) a carboxylic acid amide; and
05
06 (3) a succinimide.
07
- 08 20. The method of Claim 19 wherein the sulfur source for
09 component (1) is sulfur, hydrogen sulfide, phosphorus
10 pentasulfide, R_2S_x where R is hydrocarbyl, and x is at
11 least 2, inorganic sulfides or inorganic polysulfides,
12 thioacetamide, thiourea, mercaptans of the formula RSH
13 where R is hydrocarbyl, or a sulfur-containing
14 antioxidant.
15
- 16 21. The method of Claim 20 wherein the sulfur source for
17 component (1) is sulfur, hydrogen sulfide, phosphorus
18 pentasulfide, R_2S_x where R is C_{1-4} hydrocarbyl, and x is
19 at least 3, inorganic sulfides, or inorganic
20 polysulfides, thioacetamide, thiourea or RSH where R is
21 C_{1-40} alkyl, and the acidic molybdenum compound is
22 molybdic acid, ammonium molybdate, or alkali metal
23 molybdate.
24
- 25 22. The method of Claim 21 wherein said sulfur source is
26 sulfur, hydrogen sulfide, RSH where R is C_{1-10} alkyl,
27 phosphorus pentasulfide, or $(NH_4)_2S_{x'}$, where x' is at
28 least 1, said acidic molybdenum compound is molybdic
29 acid, or ammonium molybdate, and said basic nitrogen
30 compound is a succinimide, carboxylic acid amide, or
31 Mannich base.
32
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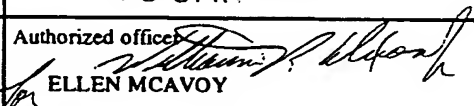
-31-

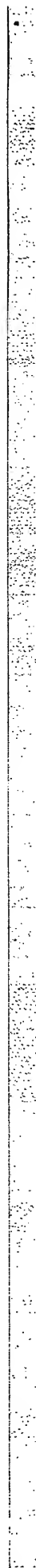
- 01 23. The method of Claim 22 wherein said basic nitrogen
02 compound is a C₂₄₋₃₅₀ hydrocarbyl succinimide,
03 carboxylic acid amide, or a Mannich base prepared from
04 a C₉₋₂₀₀ alkylphenol, formaldehyde, and an amine.
05
- 06 24. The method of Claim 23 wherein said basic nitrogen
07 compound is a polyisobutenyl succinimide prepared from
08 polyisobutenyl succinic anhydride and tetraethylene
09 pentamine or triethylene tetramine.
10
- 11 25. The method of Claim 23 wherein said basic nitrogen
12 compound is a carboxylic acid amide prepared from one
13 or more carboxylic acids of the formula R²COOH, or a
14 derivative thereof which upon reaction with an amine
15 yields a carboxylic acid amide, wherein R² is C₁₂₋₃₅₀
16 alkyl or C₁₂₋₃₅₀ alkenyl and a hydrocarbyl polyamine.
17
- 18 26. The method of Claim 25 wherein R² is C₁₂₋₂₀ alkyl or
19 C₁₂₋₂₀ alkenyl and the hydrocarbyl polyamine is
20 tetraethylene pentamine or triethylene tetramine.
21
- 22 27. The method of Claim 23 wherein said basic nitrogen
23 compound is a Mannich base prepared from dodecylphenol,
24 formaldehyde, and methylamine.
25
- 26 28. The method of Claim 23 wherein said basic nitrogen
27 compound is a Mannich base prepared from C₈₀₋₁₀₀
28 alkylphenol, formaldehyde and triethylene tetramine,
29 tetraethylene pentamine, or mixtures thereof.
30
- 31 29. The method of Claim 19 wherein the polar promoter is
32 water.
33
34

- 01 30. The method of Claim 19 wherein the carboxylic acid
02 amide of component (2) is derived from a carboxylic
03 acid of the formula R^2COOH , wherein R^2 is C_{12-20} alkyl,
04 and an ethylene amine.
05
- 06 31. The method of Claim 19 wherein the succinimide of
07 component (3) is derived from polyisobutenyl succinic
08 anhydride, wherein the polyisobutenyl group contains
09 from about 50 to 250 carbon atoms, and an ethylene
10 amine.
11
- 12 32. The method of Claim 19 wherein the lubricant
13 composition contains about 0.05 to 15% by weight of the
14 molybdenum-containing composition of component (1),
15 about 0.05 to 20% by weight of the carboxylic acid
16 amide of component (2), and about 0.05 to 15% by weight
17 of the succinimide of component (3).
18
- 19 33. The method of Claim 19 wherein the molybdenum complex
20 is reacted with the sulfur-containing compound in an
21 amount sufficient to provide about 2.0 to 4.0 atoms of
22 sulfur per atom of molybdenum.
23
- 24 34. The method of Claim 33 wherein the molybdenum complex
25 is reacted with the sulfur-containing compound in an
26 amount sufficient to provide about 2.5 to 4.0 atoms of
27 sulfur per atom of molybdenum.
28
- 29 35. The method of Claim 19 wherein the additive formulation
30 further contains a flocculant inhibitor.
31
- 32 36. The method of Claim 35 wherein the additive formulation
33 further contains a lubricity agent.
34

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US93/08471

A. CLASSIFICATION OF SUBJECT MATTER IPC(5) :C10M 159/18, 133/16, 149/00 US CL :252/42.7, 46.4, 51.5A According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) U.S. : 252/42.7, 46.4, 51.5A, 44/367, 330,331 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 4,263,152 (King et al) 21 April 1981, see col. 2, lines 10 et seq. and col. 7, lines 5-47.	1-36
Y	US, A, 3,405,064 (Miller) 08 October 1968, see col. 1, lines 19-40 and lines 55 et seq.	1-36
Y	US, A, 3,219,666 (Norman et al) 23 November 1965, see entire document.	1-36
A	US, A, 4,266,945 (Karn) 12 May 1981, see entire document.	1-36
A	US, A, 4,164,473 (Coupland et al) 14 August 1979, see entire document.	1-36
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* "A" "E" "L" "O" "P"	Special categories of cited documents: document defining the general state of the art which is not considered to be part of particular relevance earlier document published on or after the international filing date document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) document referring to an oral disclosure, use, exhibition or other means document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family
Date of the actual completion of the international search 19 NOVEMBER 1993		Date of mailing of the international search report 05 JAN 1994
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